

*Acta Cryst.* (1974). B30, 529The Crystal Structure of SnPO<sub>3</sub>F

BY ALAN F. BERNDT

Chemistry Department, University of Missouri—St. Louis, St. Louis, Missouri 63121, U.S.A.

(Received 28 September 1973; accepted 30 October 1973)

**Abstract.** Tin(II) monofluorophosphate, SnPO<sub>3</sub>F, monoclinic,  $P2_1/c$ ,  $a=4.621$  (12),  $b=12.644$  (25),  $c=6.194$  (12) Å,  $\beta=99.3$  (2)°;  $Z=4$ ,  $\rho_c=4.03$  g cm<sup>-3</sup>. The structure consists of sheets of PO<sub>3</sub>F<sup>2-</sup> ions parallel to the (100) plane with the tin(II) ions lying midway between the anion layers. Each tin(II) ion has six O and two F neighbors. Three of these O neighbors are at an average distance of 2.15 Å. For these three the O—Sn—O angle is approximately 90°.

**Introduction.** The compound SnPO<sub>3</sub>F was prepared by mixing stoichiometric amounts of SnF<sub>2</sub> and H<sub>2</sub>PO<sub>3</sub>F in a nickel crucible. The mixture was heated until all the SnF<sub>2</sub> had dissolved. After evolution of HF the white, solid residue was analyzed.

Wt. % calculated: Sn, 54.8; P, 14.3; F, 8.8; O, 22.1.  
Found: Sn, 55.1; P, 14.1; F, 8.6; O, 22.2.

The Sn, P and F analysis was by Galbraith Labs, Inc., Knoxville, Tenn. The oxygen was determined by differences.

The crystal chosen for data collection was approximately 0.1 mm in diameter and 0.5 mm in length and was mounted with the long dimension coincident with

the axis of rotation. Multiple-film equi-inclination Weissenberg data were collected for reciprocal levels  $0kl$ ,  $1kl$  and  $2kl$  with Ni-filtered Cu  $K\alpha$  radiation ( $\lambda=1.54178$  Å). Intensities were estimated visually using a calibrated intensity strip and were corrected for Lorentz and polarization effects. An absorption correction was applied assuming a cylindrical crystal with  $\mu R=3.1$  ( $\mu=622$  cm<sup>-1</sup>). No corrections were applied for anomalous dispersion or extinction. Of approximately 400 accessible reflections a total of 379 were examined. Of these 100 were unobserved.

The structure of SnPO<sub>3</sub>F is monoclinic with Laue symmetry  $2/m$ . Systematic absences of  $h0l$  reflections with  $l$  odd and  $0k0$  reflections (on a precession pattern) for  $k$  odd indicate the space group to be  $P2_1/c$ . The lattice constants were determined from a powder pattern indexed with the aid of the single-crystal data. Standard deviations in the lattice constants were estimated by a least-squares analysis.

The structure was determined by heavy-atom techniques and was refined by least-squares calculations with the *ORFLS* program (Busing, Martin & Levy, 1964) modified for an IBM 370/165 computer, to an unweighted discrepancy index,  $R=(\sum \Delta F_o/\sum |F_o|)$ , of 0.11. The minimized function was  $\sum w(F_o - F_c)^2$  with weights assigned according to:

$$w = \frac{A}{F_o} \quad \text{for } I_o \geq \sqrt{AI_{\min}}$$

$$w = \frac{I_o^2}{I_{\min}^2 F_o} \quad \text{for } I_o \leq \sqrt{AI_{\min}}$$

where  $I_{\min}=2$  and  $A$  was chosen as 20.

Separate scale factors for each observed reciprocal layer were included as refinable parameters. The largest atomic shift observed in the last least-squares cycle was  $0.006\sigma$ . The scattering factors were taken from *International Tables for X-ray Crystallography* (1962). Unobserved reflections were not included in the refinements. A three-dimensional difference electron density map calculated after the last refinement showed no extraneous peaks larger than  $2.1$  e.\*

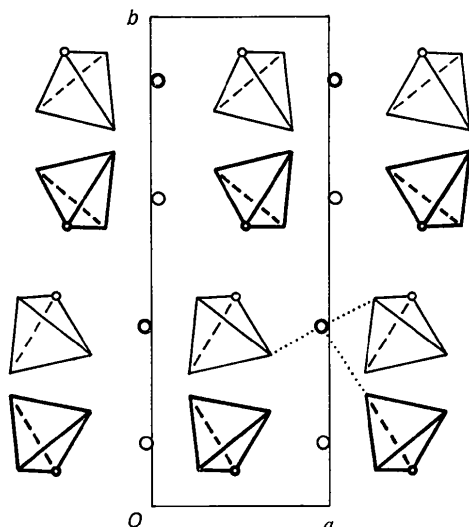


Fig. 1. Projection of the structure onto the (001) plane. The monofluorophosphate ions are illustrated as tetrahedra with the fluorine atoms shown as small circles. The tin(II) ions are illustrated as circles. The dotted lines indicate the short Sn—O distances.

\* A table of structure factors has been deposited with the British Library Lending Division as Supplementary Publication No. 30264 (3 pp.). Copies may be obtained through the Executive Secretary, International Union of Crystallography, 13 White Friars, Chester CH1 1NZ, England.

**Discussion.** The final atomic parameters and the temperature factors for each atom are given in Table 1. All atoms are in the general position. The numbers in parentheses are the standard deviations in the least significant figure as estimated from the least-squares analysis. Interatomic distances within the  $\text{PO}_3\text{F}^{2-}$  ion are shown in Table 2. The average O-P-O angle is  $114.2(35)^\circ$  while the average O-P-F angle is  $104.1(25)^\circ$ . These distances and angles are in agreement with those previously reported for the  $\text{PO}_3\text{F}^{2-}$  ion (Berndt & Sylvester, 1972).

Table 1. *Final atomic parameters*

	<i>x</i>	<i>y</i>	<i>z</i>	<i>B</i>
Sn	0.034 (2)	0.6299 (2)	0.2803 (4)	1.82 (9)
P	0.380 (6)	0.3517 (8)	0.336 (2)	1.3 (2)
O(1)	0.250 (13)	0.421 (2)	0.484 (4)	1.1 (5)
O(2)	0.205 (14)	0.270 (2)	0.202 (4)	1.6 (6)
O(3)	0.665 (16)	0.306 (3)	0.450 (5)	2.5 (7)
F	0.461 (12)	0.431 (2)	0.159 (4)	2.0 (5)

Table 2. *Interatomic distances within the monofluorophosphate ion*

P-O(1)	1.46 (4) Å	O(1)-O(2)	2.57 (4) Å
P-O(2)	1.48 (5)	O(1)-O(3)	2.44 (8)
P-O(3)	1.51 (8)	O(2)-O(3)	2.46 (8)
P-O (ave.)	1.48 (3)	O(1)-F	2.38 (5)
P-F	1.57 (3)	O(2)-F	2.39 (5)
		O(3)-F	2.47 (5)

The gross structural features of  $\text{SnPO}_3\text{F}$  (Figs. 1 and 2) are similar to those of  $\text{SnHPO}_4$  (Berndt & Lamberg, 1971) although the hydrogen bonds in the latter cannot be present in  $\text{SnPO}_3\text{F}$ . Both structures may be described as consisting of alternate layers of  $\text{PO}_3\text{F}^{2-}$  ions and tin(II) ions parallel to the (100) plane. In  $\text{SnHPO}_4$  each tin(II) ion has eight oxygen neighbors. Three of these are at an average distance of 2.26 Å. In  $\text{SnPO}_3\text{F}$  each tin(II) ion also has eight neighbors, three O atoms at an average distance of 2.15 (3) Å, three O atoms at an average distance of 3.13 Å and two F atoms at an average distance of 3.49 Å.

The three shortest Sn-O distances (illustrated by the dotted lines in Fig. 1) are associated with O atoms from three different  $\text{PO}_3\text{F}^{2-}$  ions. These three O atoms form a configuration in which the average O-Sn-O angle is  $84.9(20)^\circ$ . This feature (three short Sn-non-metal distances with angles  $\approx 90^\circ$ ) is also observed in  $\text{Sn}_3\text{PO}_4\text{F}_3$  (Berndt, 1972),  $\text{SnWO}_4$  (Jeitschko & Sleight,

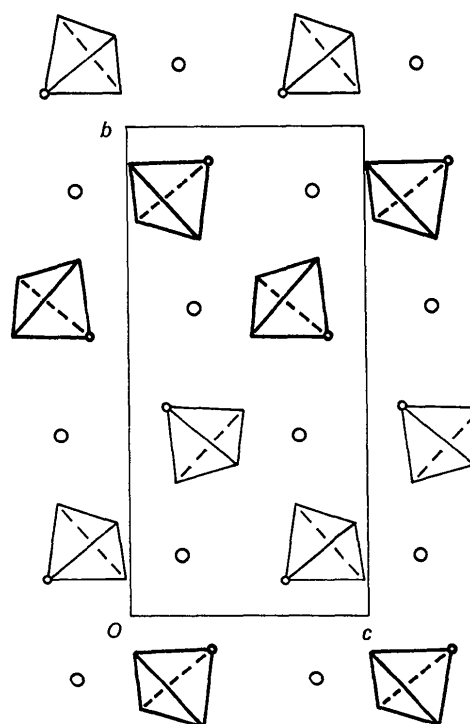


Fig. 2. Projection of the structure onto the (100) plane.

1972) and other compounds (Jeitschko & Sleight, 1972). The shortest O-O distance between O atoms not in the same  $\text{PO}_3\text{F}^{2-}$  ion or associated with the same tin(II) ion is 2.99 (7) Å.

This research received financial support from N.I.H. through grant number DE 03143-03.

#### References

- BERNDT, A. F. (1972). *J. Dent. Res.* **51**, 53-57.  
 BERNDT, A. F. & LAMBERG, R. (1971). *Acta Cryst.* **B27**, 1092-1094.  
 BERNDT, A. F. & SYLVESTER, J. M. (1972). *Acta Cryst.* **B28**, 2191-2193.  
 BUSING, W. R., MARTIN, K. O. & LEVY, H. A. (1964). *ORFLS*. Report ORNL-TM-305, Oak Ridge National Laboratory, Oak Ridge, Tennessee.  
*International Tables for X-ray Crystallography*. (1962). Vol. III, p. 212. Birmingham: Kynoch Press.  
 JEITSCHKO, W. & SLEIGHT, A. W. (1972). *Acta Cryst.* **B28**, 3174-3178.